




Field of study		Chemical Engineering					
Mode of study		stationary	Level	first cycle			
Graduate's qualification		inżynier					
Area(s) of study		nauki techniczne					
Educational profile		general academic					
Module							
Course unit		Introduction to Experimental Chemical Technology					
Code		ChEn_1A_S_C09b					
Field of specialisation							
Administering faculty		Institute of Inorganic Chemical Technology and Environmental Engineering					
ECTS		4,0	ECTS (forms)	4,0			
Form of course credit		credits	Language	english			
Electives		4	Elective group				
Form of instruction		Code	Semester	Hours	ECTS	Weight	Credit
lecture		W	4	15	1,0	0,50	credits
laboratory course		L	4	45	3,0	0,50	credits
Leading teacher		Wróbel Rafał (Rafal.Wrobel@zut.edu.pl)					
Other teachers		Bartkowiak Marcin (Marcin.Bartkowiak@zut.edu.pl), Moszyński Dariusz (Dariusz.Moszynski@zut.edu.pl), Wróbel Rafał (Rafal.Wrobel@zut.edu.pl)					
Prerequisites							
W-1		basics of chemistry					
W-2		Advanced mathematics					
W-3		basics of physics					
Module/course unit objectives							
C-1		Getting knowledge about analytical methods applied in Chemical Technology					
Course content divided into various forms of instruction						Number of hours	
T-W-1		Instrumental methods of chemical composition analysis. Selecting of a proper analytical methods. Theoretical basics of atomic spectroscopy. Inductively Coupled Plasma, ICP. Atomic absorption spectroscopy, AAS. Molecular spectra method. Infrared Spectroscopy, FTIR, UV-VIS Spectroscopy, Raman Spectroscopy RS. X-ray methods. X-Ray Fluorescence, XRF, X-Ray Microanalysis; GC and GC-MS in separation and analysis of post-reaction mixtures.					4
T-W-2		Chemical analysis of the surface of solid state. Physicochemical basics of Electro-spectroscopy methods. Methods: Electron Spectroscopy for Chemical Analysis, ESCA, including X-ray Photoelectron Spectroscopy, XPS, and Ultraviolet Photoemission Spectroscopy, UPS; Auger Electron Spectroscopy, AES, Electron Energy Loss Spectroscopy.					4
T-W-3		Adsorption/desorption methods and temperature programmed techniques. Thermogravimetry, TG, Temperature Programmed Desorption, TPD, Temperature Programmed Oxidation, TPO, Temperature Programmed Reduction, TPR, Temperature Programmed Surface Reaction, TPSR. Mass spectrometry.					3
T-W-4		Analysis of phase composition, structure and topography. X-Ray Diffraction, XRD, Reflection High Energy Electron Diffraction, RHEED, Low Energy Electron Diffraction, LEED. Mössbauer Spectroscopy. Scanning Electron Microscopy, SEM, and Transmission Electron Microscopy, TEM, Atomic Force Microscopy, AFM.					4
T-L-1		Reaction exhaust gases – MS analysis					5
T-L-2		Reactivation of catalyst for ammonia synthesis					5
T-L-3		Activation of heterogeneous catalyst based on cobalt compounds					5
T-L-4		XRD phase analysis catalysts					5
T-L-5		SEM analysis of catalysts					5
T-L-6		XRF analysis of catalysts					5
T-L-7		Deep oxidation methods in waste water treatment					5
T-L-8		GC and GC-MS methods as a methods for separation and analysis					5
T-L-9		FTIR analysis of the products					5
Student workload - forms of activity						Number of hours	
A-W-1		Obligatory attendance the lectures					15
A-W-2		Repetition of the material					15

Student workload - forms of activity						Number of hours			
A-L-1	Participation in laboratory classes					45			
A-L-2	Data evaluation and preparation of reports					45			
Teaching methods / tools									
M-1	Lecture								
M-2	Laboratory classes								
Evaluation methods (F - progressive, P - final)									
S-1	P	Written exam							
Designed learning outcomes		Reference to the learning outcomes designed for the fields of study	Reference to the learning outcomes defined for the particular areas of education	Reference to learning outcomes leading to the degree of "inżynier"	Course objectives	Course content		Teaching methods	Evaluation methods
Knowledge									
ChEn_1A_C09b_W01 Student knows theory required to design basic chemical technology process in laboratory scale.		ChEn_1A_W06 ChEn_1A_W07 ChEn_1A_W08 ChEn_1A_W11	P6S_WG_TA11		C-1	T-W-1 T-W-2	T-W-3	M-1	S-1
Skills									
ChEn_1A_C09b_U01 Student is able to mount basic laboratory scale instalation used in chemical technology and evaluate the obtained data with mother IT tools.		ChEn_1A_U01 ChEn_1A_U03 ChEn_1A_U05 ChEn_1A_U08 ChEn_1A_U09 ChEn_1A_U16	P6S_UO P6S_UU P6S_UW_TA11 P6S_UW_TA12 P6S_UW_TA14	P6S_UW_IA11 P6S_UW_IA12 P6S_UW_IA14	C-1	T-L-4 T-L-5	T-L-6	M-2	S-1
Other social / personal competences									
ChEn_1A_C09b_K01 Student is able to work in team in designing the chemical technology laboratory scale setups.		ChEn_1A_K01 ChEn_1A_K03 ChEn_1A_K04 ChEn_1A_K05	P6S_KK P6S_KO P6S_KR		C-1	T-L-1 T-L-2 T-L-3 T-L-4 T-L-5	T-L-6 T-L-7 T-L-8 T-L-9	M-1 M-2	S-1
Required reading									
1. John A. Dean, Analytical Chemistry Handbook, McGraw-Hill Companies, 2000									
1. Peter R. Griffiths, James A. de Haseth., Fourier transform infrared spectrometry, John Wiley & Sons, Hoboken, 2007, 2nd									
2. Helmut Günzler, Alex Williams, Handbook of Analytical Techniques, Wiley-VCH, 2001									
4. Paul N. Cheremisinoff, Handbook of water and wastewater treatment technology, Marcel Dekker, New York, 1995									
5. Andrzej T. Gierczycki, Łukasz Kurowski, Jan Thullie, Gas cleaning and wastewater treatment for industrial and engineering chemistry students, Wydawnictwo Politechniki Śląskiej, Gliwice, 2011									
6. Georges Guiochon, Claude L. Guillemin, Quantitative gas chromatography for laboratory analyses and on-line process control, Elsevier, Amsterdam, 1988									
7. Richard L. MacCreery, Raman spectroscopy for chemical analysis, Wiley-Interscience, New York, 2000									